#### SABANCI UNIVERSITY

Faculty of Eng. & Natural Sci.

NS-213

Basic Concepts of Physics for Scientists and Engineers

# **Instructor(s)**

Mehmet Ali Alpar

alpar@sabanciuniv.edu

http://people.sabanciuniv.edu/~alpar

FENS-2048

Teaching Assistant: Murat G. Eskin (mgeskin@sabanciuniv.edu)

### **Partial Resources:**

- A pdf version of the old NS101 notes is uploaded in SUCourse under "Resources" subsection.

- Also, the Turkish version of the NS101 notes is available (Homer Bookstore) as: "Doğayı Öğrenmek: Fizik" by M. Ali Alpar.

# **Grading Policy:**

2 midterms: 25 % each

Final: 25 %

In-class participation: 25 %

There will be NO CURVE.

#### **COURSE CONTENT**

Observing and understanding the workings of nature and expressing this understanding in models and mathematical language is fundamental to the study of science and technology. This course introduces the basic concepts of physics and the methods of modeling and solving problems in science. The subjects to be covered are (mathematical content is noted in parenthesis):

1. Mechanics: Newton's Laws of Motion. Energy Momentum and Angular Momentum. The Kepler problem. The study of systems near stable equilibrium: harmonic oscillators. Periodic motion; (the sinusoidal functions). Exponential damping and growth (the exponential function).

- 2. Statistical physics: The ideal gas law derived from mechanics. Meaning of temperature and pressure. Boltzmann definition of entropy based on the number of possible states (probability), with one simple example, the partitioning of a gas of N molecules into two half volumes: In a macroscopic system (large N), the most probable situation is much much much more probable than anything else- the 2nd Law of Thermodynamics.
- 3. Electromagnetism. Electric and magnetic fields. The concepts of flux and circulation. Maxwell's Equations and applications in the simplest geometry of two parallel plates (simple line and surface integrals). Electromagnetic wave propagation (the wave equation)
- 4. Quantum Physics. The Bohr model of the atom. Wave-particle duality and the Uncertainty Relation are needed to understand the properties of matter: What determines the size and structure of an atom? Relation between wavelength and system size. Why is there a Periodic Table? The Pauli Principle.

#### **DETAILED CONTENT**

### Mechanics:

Ch. 9: Newton's Laws of Motion

Ch. 12: Conservation of Momentum

Ch. 11: Conservation of Energy

Ch. 13: Angular Momentum

Ch. 14: Gravitation and the Kepler Problem

Ch. 16: Harmonic Oscillator

Ch. 15: Damping and the Exponential Function

### **Statistical Physics**

Ch. 17: The Ideal Gas

Ch. 10 (New): Entropy and the 2<sup>nd</sup> Law of Thermodynamics

# Electricity and Magnetization / Quantum Mechanics

Ch. 18 and Ch. 31