

**NS 218 Fundamentals of Nanoscience**  
**Spring 2022**

**Intended Audience:**

A sophomore level hands-on course for MAT, BIO, and ME majors, PHYS minors, or anyone interested in understanding phenomena governing the behavior of structures in the 1-100 nm size range.

**Objective:**

Throughout the Semester, we will make *models to crystallize our thinking* ---you know what, actually *to start thinking*. Via models we will develop intuition and heuristics to come up with conjectures/hypotheses to be tested by experiments (in wet lab and/or in silico). Accordingly, these will provide a helping hand towards having a knowledge base on how man-made nanostructures and biological nanomachines behave. At the end, we will have the insight for junior-senior level courses where both extensive measurements on materials and biological systems on all scales are made, and the formation of higher order structure is discussed.

**Moderator:**                      *Name:*                      Ali Rana Atilgan  
   *Office No.:*                FENS 2093  
   *Phone No.:*              +90 (216) 483 9525  
   *e-mail:*                    atilgan@sabanciuniv.edu  
   *URL:*                      <http://people.sabanciuniv.edu/atilgan>

**Associate:**                      Dilşah Nur Elmacı [dilsahelmaci@sabanciuniv.edu](mailto:dilsahelmaci@sabanciuniv.edu)  
**Course Data:**                   Hours/Room              Mon 17.40-19.30; Thu 18.40-19.30/FASS G018  
   *Office hours:*            TBA

**Textbook:**

Dill, K.A., Bromberg, S., and Stigter, D., *Molecular Driving Forces*, Statistical Thermodynamics in Biology. Garland Science, 2nd Ed., 2011. [QC311.5 .D55 2011](#).

**Reference:**

Israelachvili, J., *Intermolecular and Surface Forces*, 3rd Ed. Academic Press, 2011. [QD461 .I87 2011](#)

**Weeks Commencing/Topics:**

**Feb 28, Mar 7, and 14**      Act I – A story from  $1/r$  to  $1/r^6$  and empirical energy functions

Entropy as **multiplicity** and its role in nanoscience  
Coulomb's law and charge interactions  
    What do we mean by long range and how do they get weaker?  
    The **thermal energy** steps on the stage  
Dipoles – are they to stay fixed?  
Polarizability  
    Is it a material constant?  
    **Induced dipoles** – are they ubiquitously observable?  
van der Waals interactions  
Hydrogen bonds

**Mar 21, 28, and Apr 4**      Act II – Then surfaces get into the picture

Forces between particles and surfaces  
    SFA and AFM – are we getting serious, can we measure these forces?  
    Hamaker constants  
Ions take a role again – they **shield charged objects** in water  
    Debye length and Bjerrum length; any other length of similar spirit; would you suggest one?  
To what extent it is possible to unify concepts in intermolecular and inter-particle forces?  
How do similar surfaces come together in a medium?  
    Surface and interfacial energy

Apr 11                      Review and the **Midterm**

**Apr 18, 25, and May 9** Act III – Interactions lead to binding  
(**May 2** Spring Break)

Chemical kinetics – A quick recapitulation of NS 10X courses

The effect of temperature; is this thermal energy again?

How is it different from **physical kinetics**?

Binding and Adsorption Processes

The Langmuir Model – we better recollect our thermodynamics fundamentals

The Michaelis–Menten Model; yes, another old but not aged model

Sabatier's Principle – too many names floating around here! No worries just to shorten the syllabus

Delicacy – binding should be neither too tight nor too weak

**May 16, 23, and 30** Act IV – Multiple agents cooperatively in action

Self-assembly

**Benjamin Franklin spirit**

Amphiphilic molecules – what shapes of micelles do they form and why?

Biological machines

Are they different from thermodynamic cycles via which macro engines operate?

Time and length scales in the Nano-world

And "Curtain!"

June 6 Review and the **Final**

### **Class Policies**

Course will be hybrid.

**Zoom link:**

<https://sabanciuniv.zoom.us/j/98348925459>

### **Grading**

**Participation:** 10% of the final grade

**Midterm:** 40% of the final grade

**Final exam:** 50% of the final grade; covers all the material

For a passing grade, **need to collect 45% before participation** added.