

**NS 218 Fundamentals of Nanoscience**  
**Spring 2023**

**Intended Audience:**

A sophomore level hands-on course for MAT, BIO, and ME majors, PHYS minors, or anyone interested in understanding phenomena governing the behavior of structures in the 1-100 nm size range.

**Objective:**

Throughout the Semester, we will make *models to crystallize our thinking* ---you know what, actually *to start thinking*. Via models we will develop intuition and heuristics to come up with conjectures/hypotheses to be tested by experiments (in wet lab and/or in silico). Accordingly, these will provide a helping hand towards having a knowledge base on how man-made nanostructures and biological nanomachines behave. At the end, we will have the insight for junior-senior level courses where both extensive measurements on materials and biological systems on all scales are made, and the formation of higher order structure is discussed.

**Moderator:**

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**Associate:**

**Course Data:**

<u>Hours/Room</u>	Mon 11.40-12.30; Tue 12.40-14.30/FENS G029
Office hours:	TBA
<u>Zoom link:</u>	<a href="https://sabanciuniv.zoom.us/j/97433865395">https://sabanciuniv.zoom.us/j/97433865395</a>

**Textbook:**

Dill, K.A., Bromberg, S., and Stigter, D., *Molecular Driving Forces*, Statistical Thermodynamics in Biology. Garland Science, 2nd Ed., 2011. [QC311.5 .D55 2011](#).

**Reference:**

Israelachvili, J., *Intermolecular and Surface Forces*, 3rd Ed. Academic Press, 2011. [QD461 .I87 2011](#)

**Weeks Commencing/Topics:**

**Feb 27, Mar 6, and 13** Act I – A story from  $1/r$  to  $1/r^6$  and empirical energy functions

Entropy as **multiplicity** and its role in nanoscience

Coulomb's law and charge interactions

What do we mean by long range and how do they get weaker?

The **thermal energy** steps on the stage

Dipoles – are they to stay fixed?

Polarizability

Is it a material constant?

**Induced dipoles** – are they ubiquitously observable?

van der Waals interactions

Hydrogen bonds

**Mar 20, 27, and Apr 3** Act II – Then surfaces get into the picture

Forces between particles and surfaces

SFA and AFM – are we getting serious, can we measure these forces?

Hamaker constants

Ions take a role again – they **shield charged objects** in water

Debye length and Bjerrum length; any other length of similar spirit; would you suggest one?

To what extent it is possible to unify concepts in intermolecular and inter-particle forces?

How do similar surfaces come together in a medium?

Surface and interfacial energy

**Apr 10**

**Exam I**

**Apr 17, 24, and May 2** Act III – Interactions lead to binding  
**May 1** International Workers' Day (National holiday)

Chemical kinetics – A quick recapitulation of NS 10X courses  
The effect of temperature; is this thermal energy again?  
How is it different from **physical kinetics**?

Binding and Adsorption Processes

The Langmuir Model – we better recollect our thermodynamics fundamentals  
The Michaelis–Menten Model; yes, another old but not aged model  
Sabatier's Principle – too many names floating around here! No worries just to shorten the syllabus  
Delicacy – binding should be neither too tight nor too weak

**May 8 and 15** Act IV – Multiple agents cooperatively in action

Self-assembly

**Benjamin Franklin spirit**  
Amphiphilic molecules – what shapes of micelles do they form and why?

**May 22** **Exam II**

**May 29**

Biological machines

Are they different from thermodynamic cycles via which macro engines operate?

Time and length scales in the Nano-world

And "Curtain!"

### **Class Policies**

It will be the students' decision to follow **online lectures**, held in classrooms, **via Zoom or in the classroom**.

**Zoom link:** <https://sabanciuniv.zoom.us/j/97433865395>

*"We may have to revise the course plan according to the countrywide reassessment to be made regarding higher education. This is expected to happen at the beginning of April. The **content** to be delivered is **certain** but the **method of course delivery, the number and dates of exams**, and some other details are **subject to change**."*

### **Grading**

Exam I and II: 30% each, total is for 60% of the final grade

Final exam: 40% of the final grade; covers all the material